

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Application No. :

U.S. National Serial No. :

Filed :

PCT International Application No. : PCT/FR2003/003205

VERIFICATION OF A TRANSLATION

I, Charles Edward SITCH BA,

Deputy Managing Director of RWS Group Ltd UK Translation Division, of Europa House, Marham Way, Gerrards Cross, Buckinghamshire, England declare:

That the translator responsible for the attached translation is knowledgeable in the French language in which the below identified international application was filed, and that, to the best of RWS Group Ltd knowledge and belief, the English translation of the amended sheets of the international application No. PCT/FR2003/003205 is a true and complete translation of the amended sheets of the above identified international application as filed.

I hereby declare that all the statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the patent application issued thereon.

Date: April 21, 2005

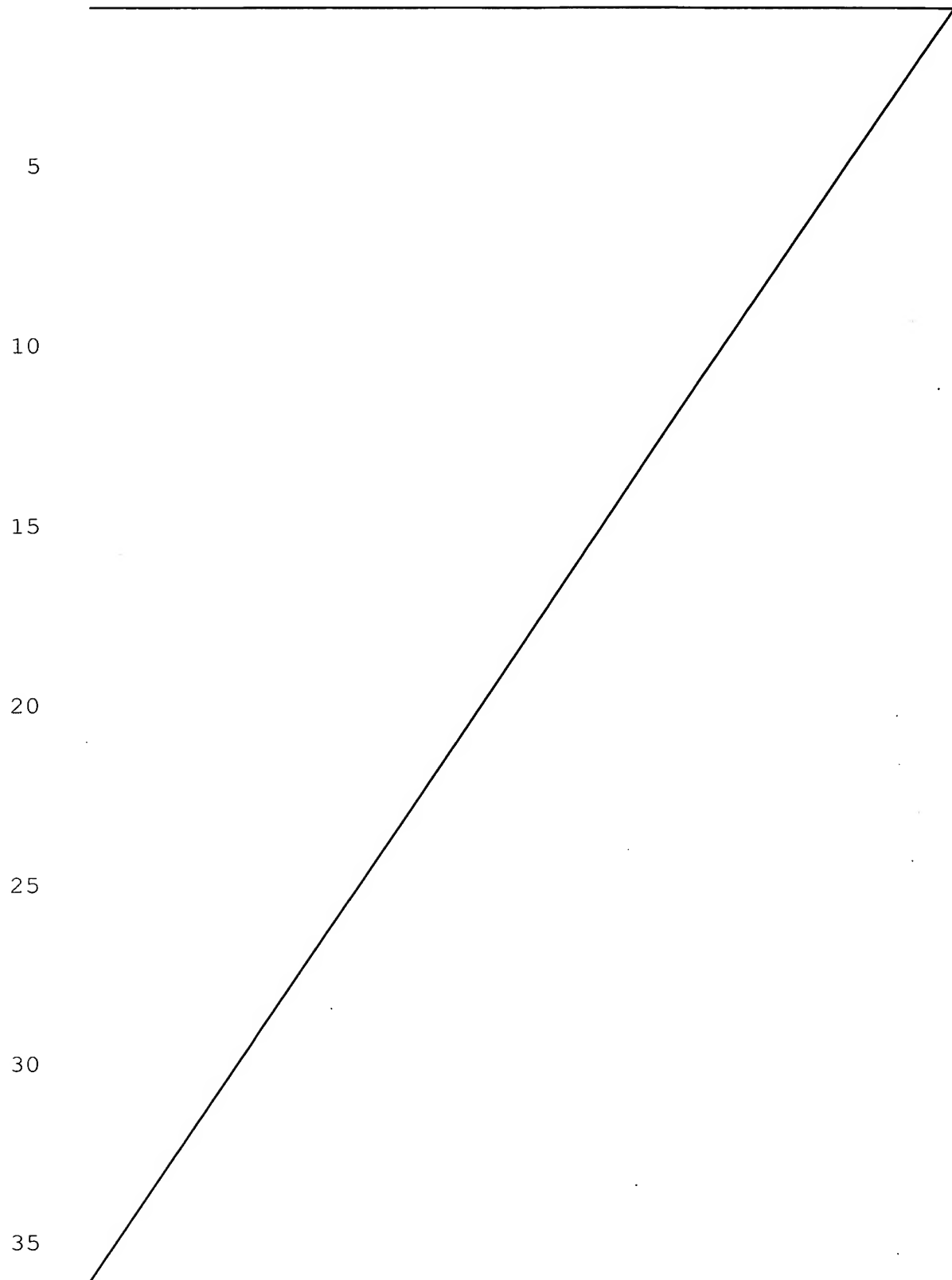
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For and on behalf of RWS Group Ltd

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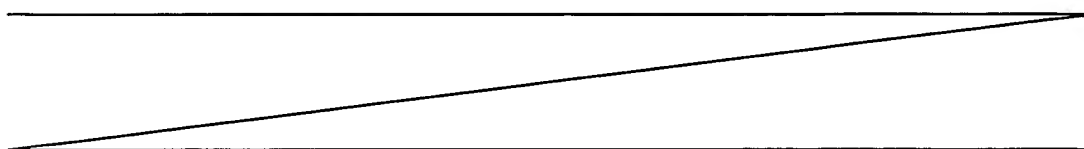
in which R1, R2, R3, R4 and R5, which are identical to

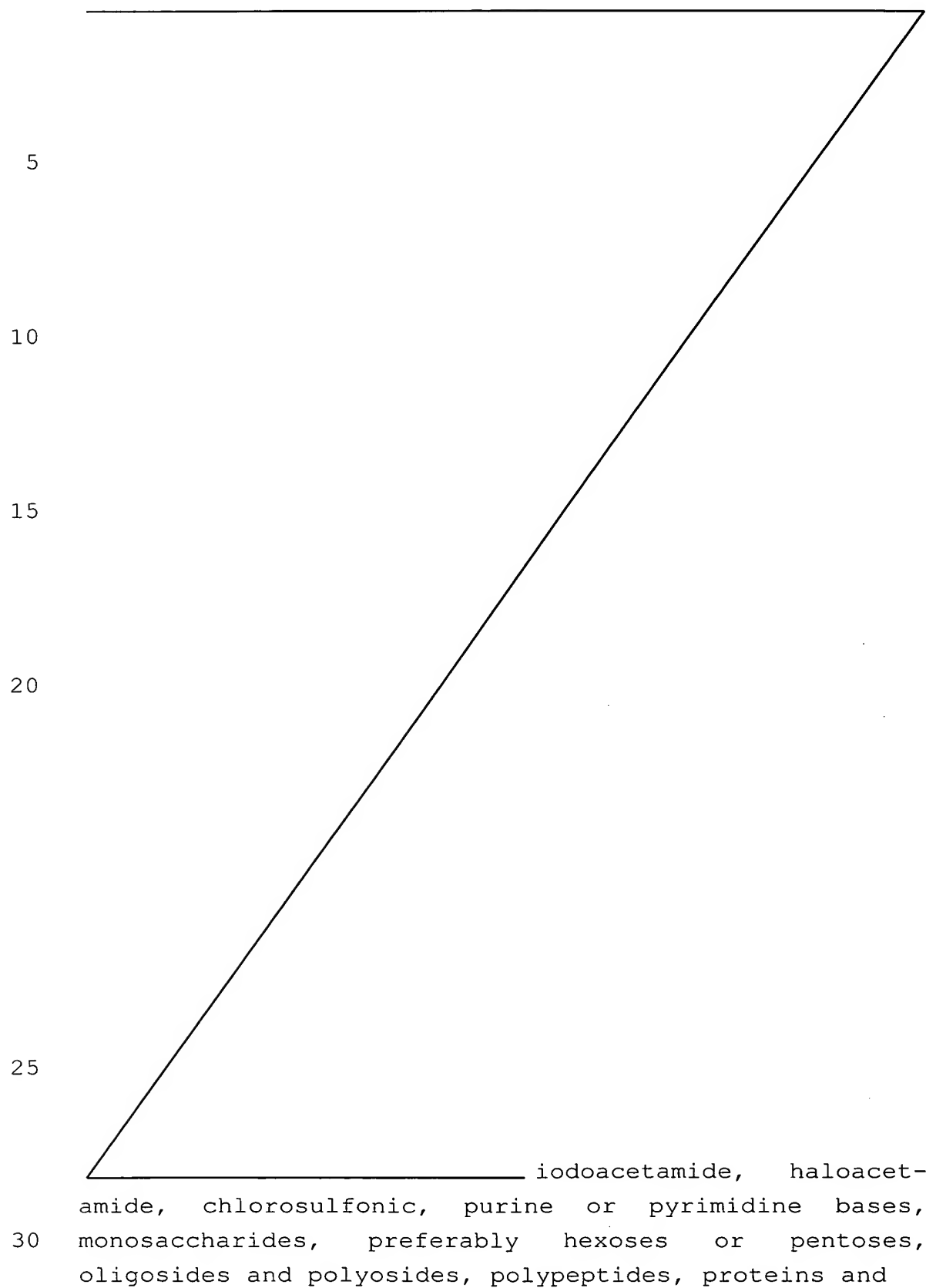
AMENDED SHEET

or different from one another, are chosen from the group comprising the following radicals or groups: hydrogen, hydroxyl, halogen, acetyl, amino, phosphate, nitro, sulfonate, carboxyl, alkylcarboxyl having from 2
5 to 30 carbon atoms, alkyl having from 1 to 30 carbon atoms, cycloalkyl having from 3 to 12 carbon atoms, alkyloxy having from 1 to 30 carbon atoms, haloalkyl having from 1 to 30 carbon atoms, hydroxyalkyl having from 1 to 30 carbon atoms, alkyl ester having from 2 to
10 40 carbon atoms, nitroalkyl having from 1 to 30 carbon atoms, carboxyalkyl having from 2 to 30 carbon atoms, aminoalkyl having from 1 to 30 carbon atoms, sulfoalkyl having from 1 to 30 carbon atoms, aryl, aryloxy, arylalkyl, haloaryl, aryl ester, succinimidyl ester,
15 isothiocyanate, maleimide, iodoacetamide, haloacetamide, chlorosulfonic, purine or pyrimidine bases, monosaccharides, preferably hexoses or pentoses, oligosides and polyosides, polypeptides, proteins and phospholipids,
20 R3 and R5 not each representing hydrogen when R1 represents a group $-\text{CH}_2-\text{CH}_2-\text{COOH}$, R2 represents a hydroxyl group and R4 represents a group $-\text{COOH}$, these phthaleins containing no more than 1% by weight, preferably no more than 0.5% by weight, and even more
25 preferably not more than 0.2% by weight, of residual impurities.

A phthalein that is particularly advantageous, in particular for ophthalmic applications, is fluorescein
30 having such a purity.

It is known practice to prepare the phthaleins of formula (I) by condensation of a phthalic anhydride derivative and of a
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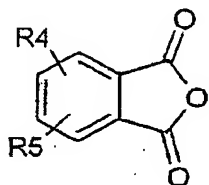




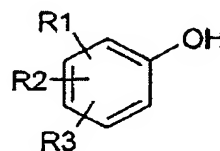
phospholipids,

R3 and R5 are not hydrogen when R1 is a group $-\text{CH}_2-\text{CH}_2-\text{COOH}$, R2 is a hydroxyl group and R4 is a group $-\text{COOH}$,

- 5 by condensation of a phthalic anhydride derivative of formula (II) with a phenol or naphthol compound of formula (III)



(II)



(III)

- 10 in which R1, R2, R3, R4 and R5 have the same meanings as above,
in a solvent consisting of an organic acid ester.

- Particularly advantageously, the starting compound
15 (III), which is condensed with the phthalic anhydride (II), is chosen from the group comprising in particular resorcinol, orcinol, naphthol, pyrogallol, alkylaminophenol and arylaminophenol.

- 20 When resorcinol is used as starting product, the method in accordance with the invention makes it possible to prepare fluorescein.

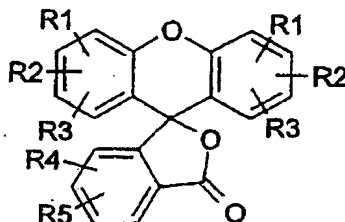
Advantageously, the solvent used in the method

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CLAIMS

1. A phthalein of general formula (I):

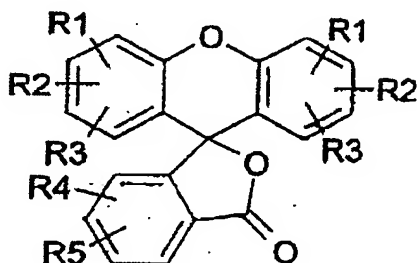


(I)

5 in which R1, R2, R3, R4 and R5, which are identical to or different from one another, are chosen from the group comprising the following radicals or groups:
10 hydrogen, hydroxyl, halogen, acetyl, amino, phosphate, nitro, sulfonate, carboxyl, alkylcarboxyl having from 2 to 30 carbon atoms, alkyl having from 1 to 30 carbon atoms, cycloalkyl having from 3 to 12 carbon atoms, alkyloxy having from 1 to 30 carbon atoms, haloalkyl having from 1 to 30 carbon atoms, hydroxyalkyl having
15 from 1 to 30 carbon atoms, alkyl ester having from 2 to 40 carbon atoms, nitroalkyl having from 1 to 30 carbon atoms, carboxyalkyl having from 2 to 30 carbon atoms, aminoalkyl having from 1 to 30 carbon atoms, sulfoalkyl having from 1 to 30 carbon atoms, aryl, aryloxy,
20 arylalkyl, haloaryl, aryl ester, succinimidyl ester, isothiocyanate, maleimide, iodoacetamide, haloacetamide, chlorosulfonic, purine or pyrimidine bases, monosaccharides, preferably hexoses or pentoses, oligosides and polyosides, polypeptides, proteins and
25 phospholipids,
R3 and R5 are not hydrogen when R1 is a group -CH₂-CH₂-COOH, R2 is a hydroxyl group and R4 is a group -COOH,
these phthaleins containing no more than 1% by weight,
30 preferably no more than 0.5% by weight, and even more preferably not more than 0.2% by weight, of residual impurities.

2. The phthalein as claimed in claim 1 consisting of fluorescein.

3. A method for preparing phthaleins, from which the residual impurities have been removed, having the general formula (I):

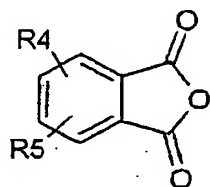


(I)

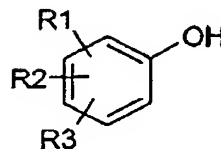
in which R1, R2, R3, R4 and R5, which are identical to
10 or different from one another, are chosen from the
group comprising the following radicals or groups:
hydrogen, hydroxyl, halogen, acetyl, amino, phosphate,
nitro, sulfonate, carboxyl, alkylcarboxyl having from 2
to 30 carbon atoms, alkyl having from 1 to 30 carbon
15 atoms, cycloalkyl having from 3 to 12 carbon atoms,
alkyloxy having from 1 to 30 carbon atoms, haloalkyl
having from 1 to 30 carbon atoms, hydroxyalkyl having
from 1 to 30 carbon atoms, alkyl ester having from 2 to
40 carbon atoms, nitroalkyl having from 1 to 30 carbon
20 atoms, carboxyalkyl having from 2 to 30 carbon atoms,
aminoalkyl having from 1 to 30 carbon atoms, sulfoalkyl
having from 1 to 30 carbon atoms, aryl, aryloxy,
arylalkyl, haloaryl, aryl ester, succinimidyl ester,
isothiocyanate, maleimide, iodoacetamide, haloacet-
25 amide, chlorosulfonic, purine or pyrimidine bases,
monosaccharides, preferably hexoses or pentoses,
oligosides and polysides, polypeptides, proteins and
phospholipids,
R3 and R5 are not hydrogen when R1 is a
30 group -CH₂-CH₂-COOH, R2 is a hydroxyl group and R4 is a

group -COOH,
characterized in that a phthalic anhydride derivative
of formula (II) is condensed with a phenol or naphthol
compound of formula (III)

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(II)



(III)

in which R1, R2, R3, R4 and R5 have the same meanings
as above,
10 the condensation being carried out in a solvent
consisting of an organic acid ester.

4. The method as claimed in claim 3, in which the
compound of formula (III) is chosen from the group
15 comprising resorcinol, orcinol, naphthol, pyrogallol,
alkylaminophenol and arylaminophenol.

5. The method as claimed in either of claims 3 and 4,
in which the solvent is an organic acid ester of
20 formula (IV)



in which R₆ is chosen from the group comprising the
25 following radicals or groups: hydrogen, alkyl having
from 1 to 30 carbon atoms, cycloalkyl having from 3 to
12 carbon atoms, haloalkyl having from 1 to 30 carbon
atoms, hydroxyalkyl having from 1 to 30 carbon atoms,
nitroalkyl having from 1 to 30 carbon atoms, aryl,
30 aryloxy, alkylaryl, arylalkyl, substituted arylalkyl,
haloaryl, aryl ester, alkyl ester having from 2 to 40
carbon atoms, and alkyloxy having from 1 to 30 carbon

atoms, R₇ representing one of the following groups:
alkyl having from 1 to 30 carbon atoms, cycloalkyl
having from 3 to 12 carbon atoms, haloalkyl having from
1 to 30 carbon atoms, hydroxyalkyl having from 1 to 30
5 carbon atoms, nitroalkyl having from 1 to 30 carbon
atoms, aryl, aryloxy, alkylaryl, arylalkyl, substituted
arylalkyl, haloaryl, aryl ester, alkyl ester having
from 2 to 40 carbon atoms, or alkyloxy having from 1 to
30 carbon atoms.

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6. The method as claimed in one of claims 3 to 5,
characterized in that the organic acid ester is chosen
from the group comprising methyl, ethyl, propyl or
butyl benzoate, methyl, ethyl, propyl or butyl
15 heptanoate, methyl, ethyl, propyl or butyl octanoate,
methyl, ethyl, propyl or butyl laurate, methyl, ethyl,
propyl or butyl myristate or methyl, ethyl, propyl or
butyl palmitate, and mixtures thereof.

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7. The method as claimed in one of claims 3 to 6,
characterized in that the condensation reaction is
carried out at between 150°C and 250°C, optionally
under pressure.

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8. The method as claimed in one of claims 3 to 7,
characterized in that it is carried out in the presence
of a catalyst chosen from the group comprising in
particular Lewis acids, such as ZnCl₂ or AlCl₃, Brönsted
acids such as H₂SO₄ or polyphosphoric acid, preferably
30 an alkali metal hydrogen sulfate, and more preferably
potassium hydrogen sulfate (KHSO₄) or sodium hydrogen
sulfate (NaHSO₄).

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9. A method for acidifying the product resulting from
the condensation of a phthalic anhydride derivative of
formula (II) with a phenol or naphthol compound of
formula (III), the formulae (II) and (III) being those
of claim 3, characterized in that it is carried out in

an anhydrous organic medium, by the addition of a strong acid or one of its precursors, chosen from the group comprising sulfuric acid, hydrochloric acid, hydrobromic acid, hydrofluoric acid, hydriodic acid, polyphosphoric acid, pyrophosphate (P_2O_5), and mixtures thereof, the acidification being carried out until the phthalein crystals resulting from the condensation are converted to phthalein crystals having a different structure.

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10. The method as claimed in claim 9, characterized in that the condensation product is the product obtained by the method as claimed in any one of claims 3 to 8.

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11. The method as claimed in claim 9 or 10, characterized in that it comprises a step consisting in washing the product obtained after acidification, said washing step being carried out with a washing solution chosen from the group comprising water, alcohols, ketones, ethers and halogenated solvents, pure or as a mixture, until the crystals are reconverted to the structure that they had before the acidification reaction.

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12. A method for preparing a fluorescein having a purity such that its content of each of the by-products of the reaction is less than or equal to 0.2%, and preferably less than or equal to 0.1%, the sum of the contents of each of these by-products being less than or equal to 0.5%, said method comprising the following successive steps:

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- condensing phthalic anhydride with resorcinol, in a solvent consisting of an ester of an aliphatic or aromatic organic acid, preferably ethyl or methyl benzoate or ethyl or methyl palmitate, in the presence of a catalyst chosen from the group comprising in particular Lewis acids or Brönsted acids, and preferably an alkali metal hydrogen

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sulfate,

- suspending the red-colored crystals obtained in the preceding step in an anhydrous solvent chosen from the group comprising alcohols such as absolute ethanol, ketones such as acetone, ethers, halogenated solvents, or mixtures thereof,
- acidifying the suspension thus obtained by the addition of a strong acid or one of its precursors, chosen from the group comprising in particular sulfuric acid, hydrochloric acid, hydrobromic acid, hydrofluoric acid, hydriodic acid, polyphosphoric acid, pyrophosphate (P_2O_5), and mixtures thereof, until the red-colored crystals are converted to yellow-colored crystals exhibiting the X-ray diffraction analysis of figure 2,
- washing the crystals obtained with a washing solution chosen from the group comprising water, alcohols, ketones, ethers and halogenated solvents, pure or as a mixture, this washing being continued until the yellow-colored crystals are reconverted to red-colored crystals.

13. The method as claimed in one of claims 9 to 12, characterized in that the acidification is carried out by sparging gaseous hydrochloric acid into the phthalein suspension or by the action, on this phthalein, of hydrochloric acid in solution in the anhydrous organic solvent, preferably an alcohol, a ketone, an ether or a halogenated solvent, used alone or as a mixture, even more preferably isopropanol, absolute ethanol or acetone, pure or as a mixture.

14. The method as claimed in one of claims 3 to 13, characterized in that the catalyst used for the condensation reaction consists of the hydrogen sulfate of an alkali metal, preferably potassium hydrogen sulfate or sodium hydrogen sulfate.

15. A yellow-colored fluorescein crystal having the X-ray diffraction analysis of figure 2.

5 16. A yellow-colored 4',5'-dimethylfluorescein crystal having the X-ray diffraction analysis of figure 4.

17. A reddish-brown- or mahogany-colored 4',5'-dihydroxyfluorescein crystal having the X-ray diffraction analysis of figure 6.

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18. A phthalein obtained by means of the method as claimed in any one of claims 3 to 14.

15 19. A fluorescein obtained by means of a method as claimed in any one of claims 3 to 14.

20. A 4',5'-dimethylfluorescein obtained by means of a method as claimed in any one of claims 3 to 14.

20 21. A 4',5'-dihydroxyfluorescein obtained by means of a method as claimed in any one of claims 3 to 14.

22. The use of the fluorescein as claimed in claim 2 or obtained according to the method of claims 3 to 14, 25 in pharmaceutical applications in diagnosis, especially in medical imaging, or in the field of biotechnological applications.